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Key indicators

Single-crystal X-ray study
 $T = 150$ K
 Mean $\sigma(C-C) = 0.004$ Å
 R factor = 0.020
 wR factor = 0.045
 Data-to-parameter ratio = 15.5

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1-Isopropyl-3-(2-morpholinioethyl)benzimidazolium diiodide

The title compound, $C_{16}H_{25}N_3O^{2+} \cdot 2I^-$, was synthesized from 1-(2-morpholinoethyl)benzimidazole and isopropyl iodide in tetrahydrofuran. In the molecule, the benzimidazole ring is connected to the morpholine ring by an ethylene group. The crystal structure has been determined at 150 K and exhibits intermolecular $C-H \cdots I$ interactions.

Received 7 October 2004

Accepted 20 October 2004

Online 30 October 2004

Comment

For some years we have synthesized and investigated crystal structures of many benzimidazole derivatives, which constitute an important class of heterocyclic compounds (Akkurt *et al.*, 2003, 2004, 2004*a,b*; Öztürk *et al.*, 2001, 2003; Türktekin *et al.*, 2004). They also show versatile pharmacological activities, such as antibacterial, antifungal, antihelmintic, antiallergic, antineoplastic, local analgesic, antihistaminic, vasodilative, hypotensive and spasmolytic activities (Easmon *et al.*, 2001; Güneş & Coşar, 1992). We have also observed that many benzimidazole derivatives and related heterocyclic compounds have shown considerable antimicrobial activities against standard strains: *Enterococcus faecalis* (ATCC 29212), *Staphylococcus aureus* (ATCC 29213), *Escherichia coli* (ATCC 25922), *Pseudomonas aeruginosa* (ATCC 27853) and yeasts *Candida albicans* and *Candida tropicalis* (Küçükbay *et al.*, 2001, 2003, 2004). The aim of this study was to synthesize and elucidate the crystal structure of the new benzimidazole title compound, (I).

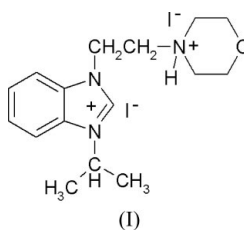


Fig. 1 shows the molecular structure of (I) and the atomic numbering scheme. Selected geometric parameters are listed in Table 1. All bond distances and angles lie within the ranges of normally accepted values.

In (I), the benzimidazole ring (N2/C7–C12/N3/C13) is essentially planar, with maximum deviations of 0.009 (2) Å for N2 and -0.013 (2) Å for C13. The morpholine ring (O1/C2/C1/N1/C4/C3) has a chair conformation (Boeyens, 1978), and puckering parameters $Q_T = 0.575$ (2) Å, $\theta = 0.0$ (2)° and $\varphi = 34$ (5)° (Cremer & Pople, 1975).

The crystal structure is stabilized by van der Waals interactions; close intermolecular contacts are listed in Table 2. The molecular packing and hydrogen-bonding interactions are illustrated in Fig. 2.

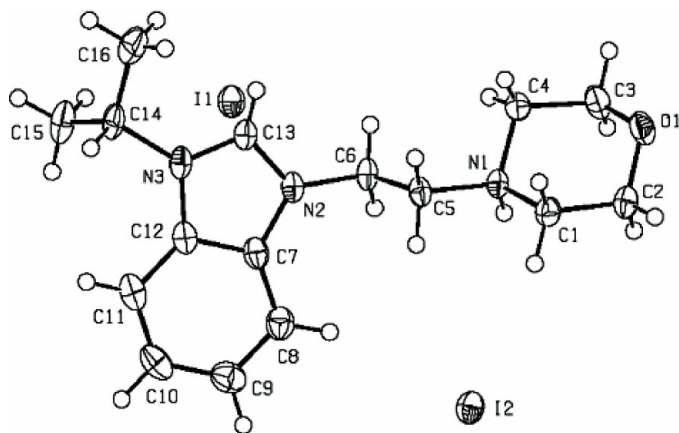


Figure 1
An ORTEP-3 plot (Farrugia, 1997) of the title compound, showing the atom-numbering scheme and 50% probability displacement ellipsoids.

Experimental

1-(2-Morpholinoethyl)benzimidazole was synthesized from benzimidazole and *N*-(2-chloroethyl)morpholine hydrochloride according to the literature method of Akkurt *et al.* (2004). A mixture of 1-(2-morpholinoethyl)benzimidazole (13.04 g, 56.45 mmol) and isopropyl iodide (11.28 ml, 112.90 mmol) was heated on a water bath for 3 h. The mixture was cooled to room temperature and Et₂O (20 ml) was added to precipitate the crude product. The crude product was then crystallized from EtOH/Et₂O (3:1) mixture (yield: 17.2 g, 58%; m.p.: 531–532 K). ¹H NMR (D₂O): δ 1.6 [d, CH(CH₃)₂, 6H], 3.4 (t, CH₂CH₂-morpholine, 2H), 3.8 (t, ring methylene, 4H), 3.9 (t, CH₂CH₂-morpholine, 2H), 4.8–5.1 (m, CHMe₂, 1H), 4.9 (t, ring methylene, 4H), 7.6–8.2 (m, Ar-H, 4H), 9.5 (s, 2-CH, 1H). Analysis calculated for C₁₆H₂₅N₃O²⁺: C 36.29, H 4.72, N 7.94%; found: C 37.23, H 4.66, N 7.27%.

Crystal data

C ₁₆ H ₂₅ N ₃ O ²⁺ ·2I ⁻	<i>D</i> _x = 1.754 Mg m ⁻³
<i>M</i> _r = 529.19	Mo <i>K</i> α radiation
Monoclinic, <i>P</i> ₂ ₁ / <i>c</i>	Cell parameters from 4269 reflections
<i>a</i> = 12.078 (5) Å	<i>θ</i> = 1.7–27.2°
<i>b</i> = 19.923 (5) Å	<i>μ</i> = 3.14 mm ⁻¹
<i>c</i> = 8.336 (5) Å	<i>T</i> = 150 K
<i>β</i> = 92.537 (5)°	Block, colorless
<i>V</i> = 2003.9 (15) Å ³	0.40 × 0.36 × 0.30 mm
<i>Z</i> = 4	

Data collection

Stoe IPDS-II diffractometer	4269 independent reflections
<i>ω</i> scans	3736 reflections with <i>I</i> > 2σ(<i>I</i>)
Absorption correction: by integration (<i>X-RED32</i>) (Stoe & Cie, 2002)	<i>R</i> _{int} = 0.032
<i>T</i> _{min} = 0.366, <i>T</i> _{max} = 0.452	<i>θ</i> _{max} = 27.1°
14805 measured reflections	<i>h</i> = -15 → 15
	<i>k</i> = -25 → 25
	<i>l</i> = -10 → 10

Refinement

Refinement on <i>F</i> ²	<i>w</i> = 1/[σ ² (<i>F</i> _o ²) + (0.0287 <i>P</i>) ²]
<i>R</i> [<i>F</i> ² > 2σ(<i>F</i> ²)] = 0.020	where <i>P</i> = (<i>F</i> _o ² + 2 <i>F</i> _c ²)/3
<i>wR</i> (<i>F</i> ²) = 0.045	(Δσ) _{max} = 0.001
<i>S</i> = 0.97	Δρ _{max} = 0.54 e Å ⁻³
4269 reflections	Δρ _{min} = -0.77 e Å ⁻³
276 parameters	Extinction correction: <i>SHELXL97</i>
H atoms treated by a mixture of independent and constrained refinement	Extinction coefficient: 0.00223 (16)

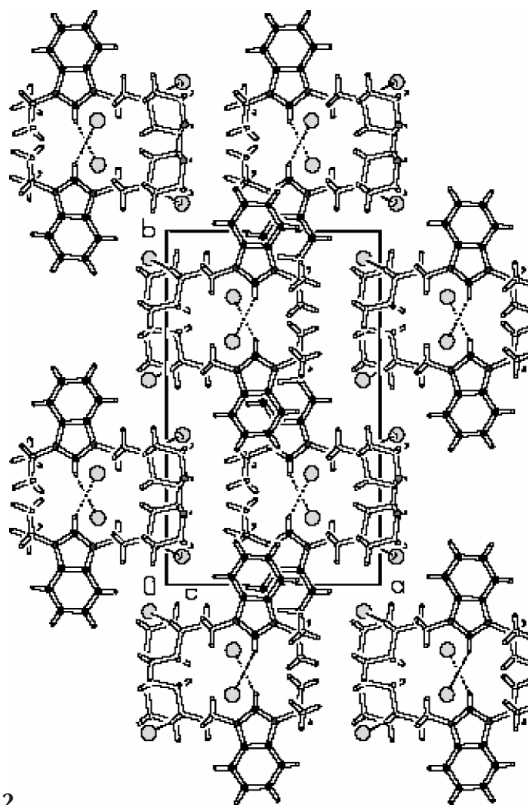


Figure 2
View of the packing and hydrogen bonds (dashed lines) of (I).

Table 1

Selected geometric parameters (Å, °).

O1–C2	1.422 (3)	N2–C7	1.387 (3)
O1–C3	1.416 (3)	N2–C13	1.333 (3)
N1–C1	1.499 (3)	N3–C12	1.395 (3)
N1–C4	1.506 (3)	N3–C13	1.325 (3)
N1–C5	1.501 (3)	N3–C14	1.488 (3)
N2–C6	1.462 (3)		
C2–O1–C3	109.72 (17)	O1–C3–C4	111.88 (19)
C1–N1–C4	109.51 (17)	N1–C4–C3	110.04 (18)
C1–N1–C5	110.33 (16)	N1–C5–C6	111.78 (17)
C4–N1–C5	113.06 (16)	N2–C6–C5	109.71 (17)
C6–N2–C7	127.20 (18)	N2–C7–C12	106.54 (18)
C6–N2–C13	124.12 (19)	N2–C7–C8	131.1 (2)
C7–N2–C13	108.27 (17)	N3–C12–C7	106.69 (18)
C12–N3–C13	108.01 (17)	N3–C12–C11	131.6 (2)
C12–N3–C14	125.33 (17)	N2–C13–N3	110.5 (2)
C13–N3–C14	126.54 (19)	N3–C14–C15	109.30 (18)
N1–C1–C2	109.59 (18)	N3–C14–C16	110.08 (19)
O1–C2–C1	111.34 (18)		
C1–N1–C5–C6	-171.71 (18)	C13–N3–C14–C16	-19.3 (3)
C4–N1–C5–C6	65.3 (2)	C13–N3–C14–C15	105.4 (3)
C7–N2–C6–C5	-82.3 (3)	C12–N3–C14–C16	165.1 (2)
C13–N2–C6–C5	106.0 (2)	N1–C5–C6–N2	176.58 (17)
C12–N3–C14–C15	-70.3 (3)		

Table 2

Hydrogen-bonding geometry (Å, °).

<i>D</i> –H··· <i>A</i>	<i>D</i> –H	H··· <i>A</i>	<i>D</i> ··· <i>A</i>	<i>D</i> –H··· <i>A</i>
N1–H1···I2 ⁱ	0.95 (3)	2.56 (3)	3.475 (3)	163 (2)
C1–H1B···I2	1.00 (3)	3.00 (3)	3.926 (3)	155.2 (19)
C13–H13···I1 ⁱⁱ	0.85 (2)	2.96 (2)	3.740 (3)	153.0 (19)

Symmetry codes: (i) 2 - *x*, -*y*, 1 - *z*; (ii) *x*₁/2 - *y*₁/2 + *z*.

The methyl H atoms were positioned geometrically, with C–H distances of 0.96 Å, and refined using a riding model, with $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{C})$. The other H atoms were located in a difference Fourier map and refined isotropically.

Data collection: *X-AREA* (Stoe & Cie, 2002); cell refinement: *X-AREA*; data reduction: *X-RED32* (Stoe & Cie, 2002); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP-3* (Farrugia, 1997); software used to prepare material for publication: *WinGX* (Farrugia, 1999).

The authors acknowledge the Faculty of Arts and Sciences, Ondokuz Mayıs University, Turkey, for the use of the Stoe IPDS-II diffractometer (purchased under grant F.279 of the University Research Fund).

References

- Akkurt, M., Öztürk, S., Küçükbay, H., Okuyucu, N. & Fun, H.-K. (2003). *Acta Cryst.* **E59**, o786–o788.
- Akkurt, M., Öztürk, S., Küçükbay, H., Orhan, E. & Büyükgüngör, O. (2004a). *Acta Cryst.* **E60**, o219–o221.
- Akkurt, M., Öztürk, S., Küçükbay, H., Orhan, E. & Büyükgüngör, O. (2004b). *Acta Cryst.* **E60**, o1263–o1265.
- Akkurt, M., Öztürk, S., Şireci, N., Küçükbay, H. & Büyükgüngör, O. (2004). *Acta Cryst.* **E60**, o1185–o1187.
- Boeyens, J. C. A. (1978). *J. Cryst. Mol. Struct.* **8**, 317–320.
- Cremer, D. & Pople, J. A. (1975). *J. Am. Chem. Soc.* **97**, 1354–1358.
- Easmon, J., Puerstinger, G., Roth, T., Fiebig, H. H., Jenny, M., Jaeger, W., Heinisch, G. & Hofmann, J. (2001). *Int. J. Cancer*, **94**, 89–96.
- Farrugia, L. J. (1997). *J. Appl. Cryst.* **30**, 565.
- Farrugia, L. J. (1999). *J. Appl. Cryst.* **32**, 837–838.
- Güneş, H. S. & Coşar, G. (1992). *Arzneim. Forsch./Drug Res.* **42**, 1045–1048.
- Küçükbay, H., Durmaz, R., Güven, M. & Günal, S. (2001). *Arzneim. Forsch./Drug Res.* **51**, 420–424.
- Küçükbay, H., Durmaz, R., Okuyucu, N., Günal, S. & Kazaz, C. (2004). *Arzneim. Forsch./Drug Res.* **54**, 64–68.
- Küçükbay, H., Durmaz, R., Orhan, E. & Günal, S. (2003). *Il Farmaco*, **58**, 431–437.
- Öztürk, S., Akkurt, M., Küçükbay, H. & Fun, H.-K. (2001). *Anal. Sci.* **17**, 1015–1016.
- Öztürk, S., Akkurt, M., Küçükbay, H., Okuyucu, N. & Fun, H.-K. (2003). *Acta Cryst.* **E59**, o1014–o1016.
- Sheldrick, G. M. (1997). *SHELXS97* and *SHELXL97*. University of Göttingen, Germany.
- Stoe & Cie (2002). *X-AREA* (Version 1.18) and *X-RED32* (Version 1.04). Stoe & Cie, Darmstadt, Germany.
- Türktekin, S., Akkurt, M., Şireci, N., Küçükbay, H. & Büyükgüngör, O. (2004). *Acta Cryst.* **E60**, o817–o819.